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Received July 1, 2004


#### Abstract

A facile approach to pyrazolo[4,3-e][1,4]diazepin-5,8-diones and pyrazolo[4,3-e]pyrrolo[1,2-a][1,4]-diazepin-5,10-diones is reported. Strategy involved the utility of $\alpha$-amino acid as a three-atom segment in the construction of diazepine skeleton on the preformed pyrazole ring.


J. Heterocyclic Chem., 42, 675 (2005).

Introduction.
Immense pharmacological importance of [1,4]benzodiazepines [1] led to a great deal of work on various facets of heteroannelated[1,4]diazepines. Thus, [1,4]diazepines fused to thiophenes [2], imidazoles [3], pyrroles [4], isoxazoles [5] and pyrazoles [6-7] were synthesized and investigated for their pharmacological activity.

Our current interest in fused pyrazoles coupled with above findings prompted us to plan the synthesis of pyra-zolo[4,3-e][1,4]diazepine derivatives. Earlier, Dewald and co-workers [6] reported the synthesis of 1,3/2,3-dialkyl-4,6-dihydro-8-aryl pyrazolo[4,3-e][1,4]diazepine-5-ones by making use of 1,3/2,3-dialkyl-4-aminopyrazolyl aryl ketones as intermediates. In the present paper, we describe the synthesis of various new fused pyrazolodiazepine derivatives through a short synthetic sequence. Utility of $\alpha$-amino acids as a three atom segment in the construction of diazepine skeleton on the preformed pyrazole ring served as a facile route to pyrazolo[4,3-e][1,4]diazepines. 1-Alkyl-4-nitro-3-n-propyl pyrazolyl-5-carboxylic acid [8] 1a/1b was chosen as the precursor.

Results and Discussion.
Reaction of 1a with thionyl chloride and subsequent condensation of the resulting acid chloride with glycine in $10 \% \mathrm{KOH}$ solution afforded 1-methyl-3-n-propyl-1H-pyrazolyl-5-carboxamido)acetic acid 2a. Treatment of 2a with hydrogen in the presence of Raney nickel at 75 psi pressure directly furnished 1-methyl-3-n-propyl-1,4,5,6,7,8-hexahydropyrazolo[4,3-e][1,4]diazepin-5,8dione 3a in $73 \%$ yield. The structural assignment of 3a was based on its ir, mass, ${ }^{1} \mathrm{H}$ and ${ }^{13} \mathrm{C} \mathrm{nmr}$ spectral data.

In the mass spectrum of $\mathbf{3 a}$ the highest peak at m/e 222 corresponds to the molecular ion. Presence of two distinct amide functions in the compound is deduced from the ir spectrum [KBr, 3178, $3070 \mathrm{~cm}^{-1}$ (two amide NH); 1689, $1669 \mathrm{~cm}^{-1}$ (amide carbonyls)]. ${ }^{1} \mathrm{H} \mathrm{nmr}$ spectrum (DMSO$\mathrm{d}_{6}$ ) displayed two amide proton signals at $\delta 10.1$ (br s, 1H)
and $\delta 8.3(\mathrm{t}, 1 \mathrm{H})$ while doublet at $\delta 3.6$ integrating for two protons is assignable to glycine $\mathrm{CH}_{2}$ group. Other signals in the ${ }^{1} \mathrm{H} \mathrm{nmr}$ spectrum are due to $N$-methyl $(\delta 4.0, \mathrm{~s}, 3 \mathrm{H})$ and $n$-propyl $\left[\delta 1.0\left(\mathrm{t}, 3 \mathrm{H}, \mathrm{CH}_{3}\right), 1.6\left(\mathrm{~m}, 2 \mathrm{H}, \mathrm{CH}_{2}\right), 2.6(\mathrm{t}\right.$, $\left.\left.2 \mathrm{H}, \mathrm{CH}_{2}\right)\right] .{ }^{13} \mathrm{C}$ nmr spectrum of $\mathbf{3 a}$ showed a total of ten signals. They include signals in the downfield region due to two carbonyl carbons ( $\delta 168.6$ and 161.7) and three carbons of pyrazole nucleus ( $\delta 148.6,125.1$ and 121.6). In the upfield region, the five signals at $\delta 13.3,21.5,26.4,38.2$ and 45.3 are due to $n$-propyl, N -methyl and N -methylene carbons. Thus, in the Raney nickel reduction of nitropyrazolylglycine derivative 2a at 75 psi pressure, initially formed aminopyrazolylglycine intermediate 4 is undergoing concomitant dehydrocylisation under the reaction conditions to provide the pyrazolodiazepine dione $\mathbf{3 a}$ in one

Scheme 1


| Compds. <br> $\mathbf{2}$ and $\mathbf{3}$ | $\mathbf{a}$ | $\mathbf{b}$ | $\mathbf{c}$ | d | e |
| :--- | :---: | :---: | :---: | :---: | :---: |
| R |  |  |  |  |  |
| $\mathrm{R}^{1}$ | $\mathrm{CH}_{3}$ | $\mathrm{CH}_{3}$ | $\mathrm{CH}_{3}$ | $\mathrm{CH}_{3}$ | $\mathrm{C}_{2} \mathrm{H}_{5}$ |
|  | H | $\mathrm{CH}_{3}$ | $\mathrm{CH}_{3}$ |  |  |

step. To verify this, compound 2a was subjected to Raney Ni reduction under mild conditions (at atmospheric pressure) and open chain intermediate 4 was isolated as the product. Compound $\mathbf{4}$ smoothly cyclised into $\mathbf{3 a}$ in the presence of 1,3-dicyclohexylcarbodiimide.
Three other $\alpha$-aminoacids, L-alanine, L-valine and Lphenylalanine furnished the corresponding pyrazolodiazepines 3b-d in good yields (Scheme-1). 1-Ethyl pyrazole derivative 1b provided the corresponding pyrazolodiazepine derivative $\mathbf{3 e}$ by participating in reaction with glycine through a similar synthetic sequence. A similar two step synthetic operation starting from $\mathbf{1 a} / \mathbf{1} \mathbf{b}$ and L-proline yielded novel fused tricyclic compounds, pyrazolo[4,3-e]-pyrrolo[1,2-a]diazepin-5,10-diones 3f-g (Scheme-2).

Condensation of 1-alkyl-4-amino-3-n-propylpyrazole-5carboxamide 5a/5b with oxalyl chloride in dichloromethane in the presence of pyridine directly yielded 1 -alkyl-3-n-propyl-1,4,5,6,7,8-hexahydropyrazolo[4,3-e]-[1,4]diazepin-5,6,8-triones 6a/6b (Scheme-3). All the pyrazolodiazepines were characterized based on their ir, mass and ${ }^{1} \mathrm{H} \mathrm{nmr}$ spectral data (Table-1).

Conclusion.
Thus, we have provided a facile new entry to pyrazolo $[4,3-e][1,4]$ diazepines. Novel fused tricyclic compounds, pyrazolo[4,3-e]pyrrolo[1,2-a]diazepines were obtained by this synthetic procedure.

Scheme 2


Table 1
Physical and Spectral Data of 2-(1-Alkyl-4-nitro-3-n-propyl-1 $H$-pyrazolyl-5-carboxamido)substituted Carboxylic Acids 2, Pyrazolodiazepindiones 3 and Pyrazolodiazepintriones 6

| Compd. <br> No. | m.p. $\left({ }^{\circ} \mathrm{C}\right)$ | Yield (\%) | ir | ${ }^{1} \mathrm{H} \mathrm{nmr}$ | $\begin{gathered} \text { MS (m/z) } \\ \left(\mathrm{M}^{+}\right) \end{gathered}$ |
| :---: | :---: | :---: | :---: | :---: | :---: |
| 2 a | 174 | 85 | $\begin{aligned} & 3246 \\ & 1725 \\ & 1654 \end{aligned}$ | $\begin{aligned} & 1.0\left(\mathrm{t}, 3 \mathrm{H}, \mathrm{CH}_{3}\right), 1.7\left(\mathrm{~m}, 2 \mathrm{H}, \mathrm{CH}_{2}\right), 2.85(\mathrm{t}, 2 \mathrm{H}, \\ & \left.\mathrm{CH}_{2}\right), 3.9\left(\mathrm{~s}, 3 \mathrm{H}, \mathrm{~N}-\mathrm{CH}_{3}\right), 4.1\left(\mathrm{~d}, 2 \mathrm{H}, \mathrm{~N}-\mathrm{CH}_{2}\right), 8.9 \\ & \text { (br s, 1H NH) } \end{aligned}$ | 270 |
| 2 b | 139 | 83 | $\begin{aligned} & 3306 \\ & 1709 \\ & 1650 \end{aligned}$ | $\begin{aligned} & 1.0\left(\mathrm{t}, 3 \mathrm{H}, \mathrm{CH}_{3}\right), 1.6\left(\mathrm{~m}, 2 \mathrm{H}, \mathrm{CH}_{2}\right), 2.4(\mathrm{~d}, 3 \mathrm{H}, \\ & \left.\mathrm{CH}_{3}\right), 2.7\left(\mathrm{t}, 2 \mathrm{H}, \mathrm{CH}_{2}\right), 4.0\left(\mathrm{~s}, 3 \mathrm{H}, \mathrm{~N}-\mathrm{CH}_{3}\right), 4.4(\mathrm{~m}, \\ & 1 \mathrm{H}, \mathrm{CH}), 8.5(\mathrm{brs} \mathrm{~s}, 1 \mathrm{H}, \mathrm{NH}) \end{aligned}$ | 284 |
| 2 c | 78 | 76 | $\begin{aligned} & 3271 \\ & 1725 \\ & 1646 \end{aligned}$ | $\begin{aligned} & 1.0\left(\mathrm{~m}, 9 \mathrm{H}, 3 \times \mathrm{CH}_{3}\right), 1.8\left(\mathrm{~m}, 3 \mathrm{H}, \mathrm{CH} \text { and } \mathrm{CH}_{2}\right) \text {, } \\ & 2.85\left(\mathrm{t}, 2 \mathrm{H}, \mathrm{CH}_{2}\right), 3.95\left(\mathrm{~s}, 3 \mathrm{H}, \mathrm{~N}-\mathrm{CH}_{3}\right), 4.7(\mathrm{~m}, 1 \mathrm{H} \text {, } \\ & \mathrm{CH}), 8.4(\mathrm{~d}, \mathrm{H}, \mathrm{NH}) \end{aligned}$ | 312 |
| 2d | 183 | 71 | $\begin{aligned} & 3268 \\ & 1703 \\ & 1654 \end{aligned}$ | $1.0\left(\mathrm{t}, 3 \mathrm{H}, \mathrm{CH}_{3}\right), 1.7\left(\mathrm{~m}, 2 \mathrm{H}, \mathrm{CH}_{2}\right), 2.8(\mathrm{t}, 2 \mathrm{H}$, $\mathrm{CH}_{2}$ ), $3.2\left(\mathrm{~m}, 2 \mathrm{H}, \mathrm{Ar}-\mathrm{CH}_{2}\right), 4.0\left(\mathrm{~s}, 3 \mathrm{H}, \mathrm{N}-\mathrm{CH}_{3}\right), 4.2$ (m, 1H, CH), 7.2 (m, 5H, Ar-H), 8.0 (br d, 1 H , NH) | 360 |
| 2 e | 160 | 81 | $\begin{aligned} & 3247 \\ & 1721 \\ & 1662 \end{aligned}$ | $0.95\left(\mathrm{t}, 3 \mathrm{H}, \mathrm{CH}_{3}\right), 1.35\left(\mathrm{t}, 3 \mathrm{H}, \mathrm{N}-\mathrm{C}-\mathrm{CH}_{3}\right), 1.6(\mathrm{~m}$, $2 \mathrm{H}, \mathrm{CH}_{2}$ ), $2.4\left(\mathrm{t}, 2 \mathrm{H}, \mathrm{CH}_{2}\right), 4.1\left(\mathrm{~d}, 2 \mathrm{H}, \mathrm{N}-\mathrm{CH}_{2}\right), 4.3$ ( $\mathrm{q}, 2 \mathrm{H}, \mathrm{N}-\mathrm{CH}_{2}$ ), $8.6(\mathrm{br} \mathrm{s}, 1 \mathrm{H}, \mathrm{NH})$ | 284 |
| 2 f | 55 | 71 | $\begin{aligned} & 1734 \\ & 1654 \end{aligned}$ | $1.0\left(\mathrm{t}, 3 \mathrm{H}, \mathrm{CH}_{3}\right), 1.8\left(\mathrm{~m}, 2 \mathrm{H}, \mathrm{CH}_{2}\right), 2.2-2.6(\mathrm{~m}, 4 \mathrm{H}$, $\mathrm{CH}_{2}$ and two pyrrolidine protons), 2.9 ( $\mathrm{m}, 2 \mathrm{H}$, $\mathrm{CH}_{2}$ ), $3.4(\mathrm{br} \mathrm{s}, 2 \mathrm{H}), 3.9\left(\mathrm{~s}, 3 \mathrm{H}, \mathrm{N}-\mathrm{CH}_{3}\right), 4.8(\mathrm{t}, 1 \mathrm{H}$, $\mathrm{C}_{5}-\mathrm{H}$ ), 8.9 (br s, $1 \mathrm{H}, \mathrm{OH}$ ) | 310 |
| 2g | 48 | 76 | $\begin{aligned} & 1725 \\ & 1651 \end{aligned}$ | $1.0\left(\mathrm{t}, 3 \mathrm{H}, \mathrm{CH}_{3}\right), 1.45\left(\mathrm{t}, 3 \mathrm{H}, \mathrm{N}-\mathrm{C}-\mathrm{CH}_{3}\right), 1.7(\mathrm{~m}$, $\left.2 \mathrm{H}, \mathrm{CH}_{2}\right), 2.1\left(\mathrm{~m}, 2 \mathrm{H}, \mathrm{CH}_{2}\right), 2.2(\mathrm{~m}, 1 \mathrm{H}), 2.4(\mathrm{~m}$, $3 \mathrm{H}, \mathrm{CH}$ and $\left.\mathrm{CH}_{2}\right), 2.8(\mathrm{~m}, 2 \mathrm{H}), 4.2\left(\mathrm{~m}, 2 \mathrm{H}, \mathrm{CH}_{2}\right)$, 4.8 (m, 1H, NH), 9.8 (br s, 1H, OH ) | 324 |

Table 1 (continued)

| Compd. No. | m.p. $\left({ }^{\circ} \mathrm{C}\right)$ | $\begin{gathered} \text { Yiel } \\ \mathrm{d} \\ (\%) \end{gathered}$ | ir | ${ }^{1} \mathrm{H} \mathrm{nmr}$ | $\begin{gathered} \mathrm{MS}(\mathrm{~m} / \mathrm{z}) \\ \left(\mathrm{M}^{+}\right) \end{gathered}$ | CHNAnalysis |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| 3a | 199 [a] | 81 | $\begin{aligned} & 3178 \\ & 3070 \\ & 1689 \\ & 1669 \end{aligned}$ | $1.0\left(\mathrm{t}, 3 \mathrm{H}, \mathrm{CH}_{3}\right), 1.6\left(\mathrm{~m}, 2 \mathrm{H}, \mathrm{CH}_{2}\right), 2.6\left(\mathrm{t}, 2 \mathrm{H}, \mathrm{CH}_{2}\right)$, 4.0 (s, 3H, N-CH $)^{2}$, 3.6 (d, 2H, N-CH ), 8.3 (br s, 1H, NH ), 10.1 (br s, 1H, NH) | 222 | $\begin{aligned} & \text { Calcd: C, } 54.04 ; \mathrm{H}, \\ & 6.35 ; \text { N, } 25.21 . \\ & \text { Found: C, } 54.19 ; \text { H, } \\ & 6.33 ; \text { N, } 25.29 . \end{aligned}$ |
| 3b | 215 [a] | 78 | $\begin{aligned} & 3326 \\ & 3155 \\ & 1699 \\ & 1669 \end{aligned}$ | $1.0\left(\mathrm{t}, 3 \mathrm{H}, \mathrm{CH}_{3}\right), 1.7\left(\mathrm{~m}, 2 \mathrm{H}, \mathrm{CH}_{2}\right), 2.4\left(\mathrm{~d}, 3 \mathrm{H}, \mathrm{CH}_{3}\right)$, $2.7\left(\mathrm{t}, 2 \mathrm{H}, \mathrm{CH}_{2}\right), 4.0\left(\mathrm{~s}, 3 \mathrm{H}, \mathrm{NCH}_{3}\right), 4.6(\mathrm{~m}, 1 \mathrm{H}$, CH ), 7.4 (br s, 1H, NH), 7.9 (br s, 1H, NH) | 236 | $\begin{aligned} & \text { Calcd: C, } 54.92 ; \mathrm{H}, \\ & \text { 6.83; N, 23.71. } \\ & \text { Found: C, 55.11; H, } \\ & \text { 6.82; N, 23.78. } \end{aligned}$ |
| 3c | 159 [a] | 75 | $\begin{aligned} & 3271 \\ & 3108 \\ & 1703 \\ & 1696 \end{aligned}$ | $1.1\left(\mathrm{~m}, 9 \mathrm{H}, 3 \times \mathrm{CH}_{3}\right), 1.7\left(\mathrm{~m}, 2 \mathrm{H}, \mathrm{CH}_{2}\right), 2.3(\mathrm{~m}, 1 \mathrm{H}$, $\mathrm{CH}), 2.9\left(\mathrm{t}, 2 \mathrm{H}, \mathrm{CH}_{2}\right), 4.0\left(\mathrm{~s}, 3 \mathrm{H}, \mathrm{N}-\mathrm{CH}_{3}\right), 4.6(\mathrm{~d}$, 1H, CH ), $8.0(\mathrm{~d}, 1 \mathrm{H}, \mathrm{NH}), 8.3(\mathrm{br} \mathrm{s}, 1 \mathrm{H}, \mathrm{NH})$ | 264 | $\begin{aligned} & \text { Calcd: C, 59.07; H, } \\ & 7.63 ; \text { N, 21.20. } \\ & \text { Found: C, 59.24; H, } \\ & 7.61 ; \text { N, 21.27. } \end{aligned}$ |
| 3d | 89 [a] | 69 | $\begin{aligned} & 3281 \\ & 3095 \\ & 1697 \\ & 1649 \end{aligned}$ | $0.9\left(\mathrm{t}, 3 \mathrm{H}, \mathrm{CH}_{3}\right), 1.7\left(\mathrm{~m}, 2 \mathrm{H}, \mathrm{CH}_{2}\right), 2.6\left(\mathrm{t}, 2 \mathrm{H}, \mathrm{CH}_{2}\right)$, 3.1 (d, 2H, Ar-CH2), 3.6 (m, 1H, CH ), 4.0 (s, 3H, N$\mathrm{CH}_{3}$ ), 7.3 (m,5H, Ar-H), 8.6 (br s, 1H, NH), 10.2 (brs, $1 \mathrm{H}, \mathrm{NH}$ ) | 312 | $\begin{aligned} & \text { Calcd: C, } 65.37 ; \mathrm{H}, \\ & \text { 6.45; N, 17.94. } \\ & \text { Found: C, 65.53; H, } \\ & \text { 6.44; N, 17.99. } \end{aligned}$ |
| 3 e | 163 [a] | 77 | $\begin{aligned} & 3324 \\ & 3153 \\ & 1694 \\ & 1666 \end{aligned}$ | $1.0\left(\mathrm{t}, 3 \mathrm{H}, \mathrm{CH}_{3}\right), 1.4\left(\mathrm{t}, 3 \mathrm{H}, \mathrm{N}-\mathrm{C}-\mathrm{CH}_{3}\right), 1.7(\mathrm{~m}, 2 \mathrm{H}$, $\mathrm{CH}_{2}$ ), $2.6\left(\mathrm{t}, 2 \mathrm{H}, \mathrm{CH}_{2}\right), 3.8\left(\mathrm{~d}, 2 \mathrm{H}, \mathrm{N}-\mathrm{CH}_{2}\right), 4.4(\mathrm{q}$, $2 \mathrm{H}, \mathrm{N}-\mathrm{CH}_{2}$ ), 7.5 (br, t, $\left.1 \mathrm{H}, \mathrm{NH}\right), 9.3(\mathrm{br} \mathrm{s}, 1 \mathrm{H}, \mathrm{NH})$ | 236 | $\begin{aligned} & \text { Calcd: C, 55.92; H, } \\ & \text { 6.83; N, 23.71. } \\ & \text { Found: C, 56.09; H, } \\ & \text { 6.83; N, 23.77. } \end{aligned}$ |
| $3 f$ | 183 [a] | 80 | $\begin{aligned} & 3160 \\ & 1684 \\ & 1653 \end{aligned}$ | $1.0\left(\mathrm{t}, 3 \mathrm{H}, \mathrm{CH}_{3}\right), 1.6\left(\mathrm{~m}, 2 \mathrm{H}, \mathrm{CH}_{2}\right), 2.6(\mathrm{~m}, 2 \mathrm{H}$, $\mathrm{CH}_{2}$ ), [2.0 ( $\mathrm{br} \mathrm{s}, 3 \mathrm{H}$ ), $2.8(\mathrm{~m}, 1 \mathrm{H}), 3.6(\mathrm{~m}, 2 \mathrm{H}), 4.1$ ( $\mathrm{m}, 1 \mathrm{H}$, due to pyrrolidine protons)], 4.0 (s, $3 \mathrm{H}, \mathrm{N}$ $\mathrm{CH}_{3}$ ), 9.4 (br s, 1H, NH) | 262 | $\begin{aligned} & \text { Calcd: C, } 59.53 ; \mathrm{H}, \\ & \text { 6.92; N, 21.36. } \\ & \text { Found: C, 59.68; H, } \\ & 6.90 ; \text { N, 21.43. } \end{aligned}$ |
| 3g | 164 [a] | 83 | $\begin{aligned} & 3169 \\ & 1683 \\ & 1644 \end{aligned}$ | $1.0\left(\mathrm{t}, 3 \mathrm{H}, \mathrm{CH}_{3}\right), 1.4\left(\mathrm{t}, 3 \mathrm{H}, \mathrm{N}-\mathrm{C}-\mathrm{CH}_{3}\right), 1.7(\mathrm{~m}, 2 \mathrm{H}$, $\mathrm{CH}_{2}$ ), $2.7\left(\mathrm{t}, 2 \mathrm{H}, \mathrm{CH}_{2}\right.$ ), [2.0 (br s, 3H ), $2.8(\mathrm{br} \mathrm{s}, 1 \mathrm{H})$, 3.6 (br s, 2H), 4.1 (br d, 1H pyrrolidine protons)], $4.4\left(\mathrm{q}, 2 \mathrm{H}, \mathrm{N}-\mathrm{CH}_{2}\right), 9.8$ (br s, 1H, NH) | 276 | $\begin{aligned} & \text { Calcd: C, } 60.85 ; \mathrm{H}, \\ & 7.30 ; \mathrm{N}, 20.28 . \\ & \text { Found: C, } 60.96 ; \mathrm{H}, \\ & 7.31 ; \mathrm{N}, 20.32 . \end{aligned}$ |
| 6 a | 297 [b] | 84 | $\begin{aligned} & 3292 \\ & 1698 \\ & 1672 \\ & 1691 \end{aligned}$ | $1.0\left(\mathrm{t}, 3 \mathrm{H}, \mathrm{CH}_{3}\right), 1.7\left(\mathrm{~m}, 2 \mathrm{H}, \mathrm{CH}_{3}\right), 2.5\left(\mathrm{t}, 2 \mathrm{H}, \mathrm{CH}_{2}\right)$, 4.0 ( $\mathrm{s}, 3 \mathrm{H}, \mathrm{N}-\mathrm{CH}_{3}$ ), 7.6 (br s, $1 \mathrm{H}, \mathrm{NH}$ ), 10.5 (br s, $1 \mathrm{H}, \mathrm{NH}$ ) | 236 | $\begin{aligned} & \text { Calcd: C, 50.84; H, } \\ & 5.12 ; \text { N, } 23.72 . \\ & \text { Found: C, 50.93, H, } \\ & 5.11 ; \text { N, 23.79. } \end{aligned}$ |
| 6 b | 265 [c] | 80 | $\begin{aligned} & 3253 \\ & 1691 \\ & 1662 \\ & 1623 \end{aligned}$ | $1.0\left(\mathrm{t}, 3 \mathrm{H}, \mathrm{CH}_{3}\right), \mathrm{I} .4\left(\mathrm{t}, 3 \mathrm{H}, \mathrm{~N}-\mathrm{C}-\mathrm{CH}_{3}\right), 1.6(\mathrm{~m}, 2 \mathrm{H},$ $\mathrm{CH}_{2}$ ), $2.4\left(\mathrm{t}, 2 \mathrm{H}, \mathrm{CH}_{2}\right), 4.3\left(\mathrm{~m}, 2 \mathrm{H}, \mathrm{N}-\mathrm{CH}_{2}\right), 7.5(\mathrm{br}$ $\mathrm{s}, 1 \mathrm{H}, \mathrm{NH}$ ), 10.1 (br s, 1H, NH) | 250 | $\begin{aligned} & \text { Calcd: C, 52.79; H, } \\ & 5.64 ; \text { N, 22.39. } \\ & \text { Found: C, 52.90; H, } \\ & 5.62 ; \text { N, 22.44. } \end{aligned}$ |

Solvent used for recrystallization: [a] ethyl acetate: $n$-hexane; [b] methanol:ethyl acetate; [c] methanol.


EXPERIMENTAL
${ }^{1} \mathrm{H}$ and ${ }^{13} \mathrm{C} \mathrm{nmr}$ spectra were recorded on a Varian Gemini (300 and 100 MHz respectively) nmr spectrometer at ambient temperature using TMS as internal standard. Mass spectrometry (70 eV) was carried out on a Perkin-Elmer Hitachi RMU-6 L
instrument. IR spectra were obtained in KBr pellets on a Shimadzu 435 instrument. Elemental Analysis was carried out on Perkin-Elmer 2400 S CHN analyzer. Melting points were determined in capillaries using Polman digital melting point apparatus (Model-mp-96). Reagents and solvents were of analytical grade. Solvents were dried before use.

2-(1-Alkyl-4-nitro-3-n-propyl-1H-pyrazolyl-5-carboxamido)substituted Carboxylic Acids (2a-g).

General Procedure.
A mixture of 1-alkyl-4-nitro-3-n-propyl pyrazolyl-5-carboxylic acid $\mathbf{1 a} / \mathbf{1 b}$ [8] ( 0.01 mol ) and thionyl chloride $(10 \mathrm{~mL})$ in benzene ( 10 mL ) was refluxed for 3 h . The reaction mixture was cooled and excess thionyl chloride was removed in vacuo. The oily residue was dissolved in benzene ( 10 mL ) and this acid chloride was cautiously added to the appropriate $\alpha$-amino acid ( 0.011 $\mathrm{mol})$ in $10 \%$ aq. KOH solution ( 20 mL ) at $10-20^{\circ} \mathrm{C}$ and reaction mixture was stirred for 15 min . The organic layer was separated
and aqueous layer was acidified with $2 N \mathrm{HCl}(10 \mathrm{~mL})$ and extracted with ethyl acetate ( $2 \times 25 \mathrm{~mL}$ ). Ethyl acetate solution was dried $\left(\mathrm{Na}_{2} \mathrm{SO}_{4}\right)$, concentrated and the resulting carboxylic acid derivative 2 was recrystallised from ethyl acetate/hexane.

1-Alkyl-6-substituted-3-n-propyl-1,4,5,6,7,8-hexahydropyrazolo-[4,3-e][1,4]diazepin-5,8-diones (3a-g).

General Procedure.
To a solution of $\mathbf{2}(5 \mathrm{mmol})$ in methanol $(25 \mathrm{~mL})$ was added Raney nickel ( 0.5 g ) and reaction mixture was placed under a hydrogen atmosphere ( 75 psi ) in a Paar hydrogenation apparatus for $4-5 \mathrm{~h}$, and then filtered through a celite bed. The catalyst was washed with methanol ( 15 mL ) and combined filtrates were evaporated to dryness in vacuo. The resulting residue was taken in $\mathrm{H}_{2} \mathrm{O}(20 \mathrm{~mL})$ and extracted with ethyl acetate ( $3 \times 25 \mathrm{~mL}$ ). The organic extracts were combined, dried $\left(\mathrm{Na}_{2} \mathrm{SO}_{4}\right)$ and evaporated under reduced pressure to give the corresponding pyrazolodiazepine dione derivative 3a-g which was recrystallised from appropriate solvent.
2-(4-Amino-1-methyl-3-n-propyl-1 H -pyrazolyl-5-carboxamido) Acetic Acid (4).

To a solution of 2-(1-methyl-4-nitro-3-n-propyl-1H-pyra-zolyl-5-carboxamido) acetic acid 2a ( $2.7 \mathrm{~g}, 5 \mathrm{mmol}$ ) in methanol ( 25 mL ), was added Raney nickel ( 0.5 g ) and reaction mixture was placed under a hydrogen atmosphere at atmospheric pressure for 4 h and then filtered through a celite bed. The catalyst was washed with methanol $(10 \mathrm{~mL})$ and the combined filtrates were evaporated to dryness in vacuo. The resulting residue was taken in water $(25 \mathrm{~mL})$ and extracted with chloroform ( $3 \times 25 \mathrm{~mL}$ ). The organic extracts were combined, dried $\left(\mathrm{Na}_{2} \mathrm{SO}_{4}\right)$ and evaporated under reduced pressure. Recrystallisation of the residue from benzene gave compound 4, yield $72 \%$, m.p. $124{ }^{\circ} \mathrm{C}$. MS: m/e $240 \mathrm{M}^{+}$; IR ( $\mathrm{KBr}, \mathrm{cm}^{-1}$ ) 3510, 3366, 3234, 1727, 1646; ${ }^{1} \mathrm{H}$ NMR ( $\mathrm{CDCl}_{3}$ ): $\delta 1.0(\mathrm{t}, 3 \mathrm{H}$, $\mathrm{CH}_{3}$ ), $1.65\left(\mathrm{~m}, 2 \mathrm{H}, \mathrm{CH}_{2}\right), 2.6\left(\mathrm{t}, 2 \mathrm{H}, \mathrm{CH}_{2}\right), 3.8-4.2(\mathrm{br} \mathrm{m}, 7 \mathrm{H}$, $\mathrm{N}-\mathrm{CH}_{3}, \mathrm{~N}-\mathrm{CH}_{2}, \mathrm{NH}_{2}$ ), 8.8 (s, $1 \mathrm{H}, \mathrm{NH}$ ).

Cyclisation of 2-(4-Amino-1-methyl-3-n-propyl-1H-pyrazolyl-5carboxamido) Acetic Acid (4).

A solution of $\mathbf{4}(0.01 \mathrm{~mol})$ and catalytic amount of 4-dimethylamino pyridine in dichloromethane ( 20 mL ) was cooled to $0^{\circ} \mathrm{C}$.

Solution of DCC ( $2.06 \mathrm{~g}, 0.01 \mathrm{~mol}$ ) in dichloromethane was added drop wise to the above solution over a period of 10 min while maintaining the temperature at $0-5{ }^{\circ} \mathrm{C}$. After the addition, the solution was stirred for an additional period of 5 min and the temperature was allowed to rise to room temperature over a period of 1 h . Separated urea derivative was filtered, washed with dichloromethane ( 15 mL ), combined organic layers were washed with water, dried $\left(\mathrm{Na}_{2} \mathrm{SO}_{4}\right)$ and evaporated in vacuo. The residue was triturated with $n$-hexane and the resulting solid was recrystallised from ethyl acetate to give pyrazolo[4,3-e][1,4]diazepin5,8 -dione derivative 3a, yield $68 \%$, m.p. $19{ }^{\circ} \mathrm{C}$.
1-Methyl/ethyl-3-n-propyl-1,4,5,6,7,8-hexahydropyrazolo[4,3-e]-[1,4]diazepin-5,6,8-trione ( $\mathbf{6 a / b}$ ).

## General Method.

To a solution of 4-amino-1-alkyl-3-n-propyl pyrazolyl-5-carboxamide $5 \mathbf{a} / \mathbf{b}(0.01 \mathrm{~mol})$ in dichloromethane $(40 \mathrm{~mL})$ and pyridine ( 2 mL ), oxalyl chloride ( $1.3 \mathrm{~g}, 0.011 \mathrm{~mol}$ ) was added drop wise and reaction mixture was stirred at room temperature for 3 h . 1-Alkyl-3-n-propyl-1,4,5,6,7,8-hexahydropyrazolo[4,3-e][1,4]-diazepin- $5,6,8$-trione $6 \mathrm{a} / \mathrm{b}$ separated out from the clear solution, which was collected by filtration, washed with water ( $2 \times 30 \mathrm{~mL}$ ) and recrystallised from suitable solvent.

## REFERENCES AND NOTES

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